# Selective Reactions using $\boldsymbol{N}$-(tert-Butoxycarbonyl)-2-(tert-butyldimethylsiloxy)pyrrole: Concise Asymmetric Syntheses of (+)-1-Deoxy-8-epi-castanospermine and its Enantiomer 

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#### Abstract

Advantage being taken of the versatility of the siloxydiene TBSOP in asymmetric synthesis, ( $6 S, 7 R, 8 S, 8 a R$ )-6,7,8-trihydroxyindolizidine 11 has been assembled from the L-threose derivative 1 in six or eight steps in $22-30 \%$ overall yield. Pivotal to the success of this total synthesis venture is the ready availability of unsaturated lactam 2 with complete stereocontrol. As a corollary, the synthesis of the known indolizidine enantiomer ent-11 confirms the feasibility of the procedure. The structure of compound 5 has been determined by X-ray crystallography.


Diastereoselective addition reactions of the heterocyclic siloxydienes 2-(trimethylsiloxy)furan (TMSOF) ${ }^{1}$ and $N$-(tert-butoxy-carbonyl)-2-(tert-butyldimethylsiloxy)pyrrole (TBSOP) ${ }^{2}$ with enantiopure aldehydo or imino precursors have been shown to be a key step in a flexible method for the construction of complex molecules bearing multiple contiguous chiral centres. In numerous ways, shown in Scheme 1, through various combinations of oxygen and nitrogen substitutions and accurate stereocontrolled manipulations, a wide variety of molecular assemblies can be constructed endowed with all-oxygen, $\mathrm{N}, \mathrm{O}-$ mixed and all-nitrogen functionalities.


Scheme 1
As part of a program directed at a general route to monosaccharides and aza-sugars of any constitution and stereochemistry, we became interested in extending this approach to functionalized indolizidines of type $\mathbf{D}$ by taking advantage of the immense versatility of TBSOP in stereocontrolled syntheses. ${ }^{2}$

We now report the successful addition of TBSOP with Land D-threose derivatives 1 and ent-1 ${ }^{3}$ and its application in concise asymmetric syntheses of ( $6 S, 7 R, 8 S, 8 \mathrm{a} R$ )-6,7,8-trihydroxyindolizidine $[(+)$-1-deoxy-8-epi-castanospermine] 11 and its enantiomer ent-11. Because of the potential value of
monocyclic and bicyclic aza-sugar compounds and alkaloids as glycoprocessing inhibitors and therapeutic agents, ${ }^{4}$ the efficient preparation of this important family of inhibitors has been of extreme interest to organic and medicinal chemists. ${ }^{5,6}$

## Results and Discussion

Synthesis.-The required protected aldehydo-threoses 1 and ent- 1 were prepared from the corresponding tartrate esters according to the literature, ${ }^{3}$ while large-scale preparation of TBSOP was achieved from pyrrole as previously reported by us. ${ }^{2}$ Optimally, compound 1 was treated with TBSOP in diethyl ether at $-80^{\circ} \mathrm{C}$ in the presence of 1.2 mol equiv. of $\mathrm{SnCl}_{4}$ (Scheme 2). The addition occurred regio- and stereo-selectively at the C- 5 carbon of TBSOP to form crystalline $\alpha, \beta$-unsaturated lactam 2 exclusively, in $80 \%$ isolated yield. The determination of the relative stereochemistry of the newly created stereocentres $\mathbf{C}(4)$ and $\mathbf{C}(5)$ was based on a single-crystal X-ray analysis of a more advanced intermediate, namely lactam 5 (vide infra), while the absolute $4 R, 5 S$ configuration was inferred from the chirality of the employed threose 1.
Our first approach to the indolizidine 11 envisaged doublebond saturation in lactam 2 to form compound 3, followed by selective removal of the N -Boc protecting group to furnish compound 4 and subsequent reduction of the lactam and ring closure to create the indolizidine skeleton. Thus, catalytic hydrogenation of lactam 2 using Pd on charcoal in NaOAcbuffered tetrahydrofuran (THF) at ambient temperature and pressure yielded saturated lactam $3(95 \%)$; however, the subsequent deprotection reaction using trimethylsilyl trifluromethanesulfonate (TMSOTf) failed to produce target compound 4 . When the reaction was performed in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at $0^{\circ} \mathrm{C}$, a single crystalline compound could be isolated in $94 \%$ yield. This product was shown spectroscopically to contain the expected amidic NH function (singlet at $\delta 6.02$ ) along with a singlet integrating to 9 H at $\delta 1.23$, presumably due to the presence of an oxygen-linked tert-butyl substituent. The structure 5 was assigned to this lactam, based on a single-crystal X-ray analysis (vide infra). This scheme was not further pursued.

Our second attempt utilized lactam 6 prepared via selective N Boc deprotection of unsaturated lactam 2 as shown in Scheme 3.
Treatment of compound 2 with TMSOTf in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ in the presence of thiophenol cleanly afforded lactam $6(75 \%)$, which


Scheme 2 Reagents and conditions: i, $\mathrm{SnCl}_{4}, \mathrm{Et}_{2} \mathrm{O},-80^{\circ} \mathrm{C}, 5 \mathrm{~h} ; \mathrm{ii}, \mathrm{H}_{2}$, $\mathrm{Pd}-\mathrm{C}, \mathrm{NaOAc}, \mathrm{THF}, 22^{\circ} \mathrm{C}, 48 \mathrm{~h}$; iii, TMSOTf, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, 0^{\circ} \mathrm{C}, 6 \mathrm{~h}$
was hydrogenated and deprotected to compound 4 ( Pd on charcoal, NaOAc, THF) in $95 \%$ isolated yield. Having the desired alcohol 4 in hand, we next examined its conversion into the indolizidine 11. We first pursued a conventional ring-closure methodology based on conversion of compound 4 into mesate 7 followed by reduction of the lactam and $\mathrm{N}-\mathrm{C}(8)$ ring closure. Selective monomesylation was thus successfully conducted by using 1.1 mol equiv. of MsCl in pyridine in the presence of a catalytic amount of 4-(dimethylamino)pyridine (DMAP) to furnish mesate 7 in $74 \%$ yield. Unfortunately, exposure of compound 7 to $\mathrm{BH}_{3} \cdot$ dimethyl sulfide (DMS) complex in THF, a reagent known to reduce selectively the lactam carbonyl
function in similar structures, ${ }^{6 t}$ met with complete failure, the unwanted $C$-nucleoside-like furanose 8 being solely generated in $45 \%$ yield with no trace of the expected indolizidine 11. One possible explanation for this competitive and preferential $\mathrm{O}(5)-\mathrm{C}(8)$ annulation might be steric factors at the $\mathrm{C}(6)$ and $\mathrm{C}(7)$ carbon atoms of the dioxolane which prevent nitrogen from attacking $C(8)$ to form a six-membered piperidine endowed with a strained trans-disposed acetonide ring.

Having thus abandoned this path, we envisaged a third plan involving prior carbonyl reduction in lactam 4, removal of the acetonide, and final NH-C(8) annulation. This route was shown to be satisfactory although some conversions were not as anticipated. As shown in Scheme 3, lactam 4 was directly exposed to an excess of $\mathrm{BH}_{3} \cdot$ DMS in THF at room temperature and the crude amine-borane adduct thus formed was subjected to acidic treatment with $2 \mathrm{~mol} \mathrm{dm}^{-3} \mathrm{HCl}$ at room temperature. Surprisingly enough, this treatment afforded isopropyl ether 9 $(71 \%)$, likely to have arisen from reduction of the lactam with concomitant regioselective opening of the acetonide at the $\mathrm{O}(7)-$ $\mathrm{C}(\mathrm{Me})_{2}$ linkage and over-reduction.

Nonetheless, amino alcohol 9 was ready for the final and crucial cyclization step. This was achieved by subjecting compound 9 to $\mathrm{PPh}_{3}-\mathrm{CCl}_{4}-\mathrm{Et}_{3} \mathrm{~N}$ in pyridine at room temperature, as suggested by Vogel. ${ }^{6 u}$ There was obtained, after ion-exchange resin purification, the 7 -isopropoxyindolizidine 10 in $79 \%$ isolated yield. This compound was recognized as such mainly through ${ }^{13} \mathrm{C}$ NMR analysis, showing the expected eleven carbon signals in the appropriate positions for two methyls, four methylenes, and five methines. Cleavage of the oxygen-carbon bond in the isopropyl ether 10 was finally accomplished, as anticipated, in a simple manner by using $\mathrm{BBr}_{3}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. This treatment afforded the indolizidine $\mathbf{1 1}$ in $67 \%$ yield, which showed ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR characteristics identical with those reported by St-Denis and Chan for its enantiomer ent -11 . ${ }^{69}$

The conversion of isopropyl ether $\mathbf{1 0}$ into triol $\mathbf{1 1}$ was also possible through a slightly longer but cleaner reaction sequence. Conventional acetylation of compound 10 ( $\mathrm{Ac}_{2} \mathrm{O}$, pyridine, DMAP) quantitatively afforded diacetate 12 , suitable for


Scheme 3 Reagents and conditions: i, PhSH, TMSOTf, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, 0^{\circ} \mathrm{C}, 1 \mathrm{~h}$; ii, $\mathrm{H}_{2}, \mathrm{Pd}-\mathrm{C}, \mathrm{NaOAc}, \mathrm{THF}, 48 \mathrm{~h}$; iii, MsCl, pyridine, DMAP ( 0.3 mol equiv.), $22^{\circ} \mathrm{C}, 7 \mathrm{~h}$; iv, $\mathrm{BH}_{3} \mathrm{DMS}\left(50 \mathrm{~mol}\right.$ equiv.), THF, $22^{\circ} \mathrm{C}, 24 \mathrm{~h}$; then $2 \mathrm{~mol} . \mathrm{dm}^{-3} \mathrm{HCl}, 20^{\circ} \mathrm{C}, 30 \mathrm{~min}$; then DOWEX $\left(\mathrm{OH}^{-}\right)$; $\mathrm{v}, \mathrm{PPh}_{3}, \mathrm{CCl}_{4}, \mathrm{Et}_{3} \mathrm{~N}$, pyridine, $20^{\circ} \mathrm{C}, 20 \mathrm{~h}$; then DOWEX $\left(\mathrm{H}^{+}\right) 2 \mathrm{~mol} \mathrm{dm}{ }^{-3} \mathrm{NH}_{4} \mathrm{OH}$; vi, $\mathrm{BBr}_{3}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 20^{\circ} \mathrm{C}, 20 \mathrm{~h}$; vii, $\mathrm{Ac}_{2} \mathrm{O}$, pyridine, DMAP ( 0.3 mol equiv.), $22^{\circ} \mathrm{C}$, 5 h ; viii, NaOMe ( 0.1 mol equiv.), $\mathrm{MeOH}, 20^{\circ} \mathrm{C}, 5 \mathrm{~h}$


$$
\begin{aligned}
& J_{5 e q .6}=5.1 \\
& J_{5 a x .6}=10.2 \\
& J_{6,7}=10.2 \\
& J_{7,8}=3.8 \\
& J_{8,8 \mathrm{a}}=1.8
\end{aligned}
$$

Fig. 1 Compound 12 in its ${ }^{82} C_{6}$ configuration. Diagnostic coupling constants ( Hz ) and NOE interactions (arrows).
accurate structural and conformational ${ }^{1} \mathrm{H}$ NMR analyses. The study allowed complete assignments of all the proton resonances as indicated in Fig. 1, and was consistent with the piperidine ring adopting a ${ }^{8 a} C_{6}$ conformation in solution with no significant distortion to accommodate the trans-fused fivemembered heterocycle.

Selective removal of the $7-O$-isopropyl group in compound 12 was smoothly accomplished by acidic treatment $\left(\mathrm{BBr}_{3}\right)$ to furnish diacetate $13(95 \%)$, which was converted into triol 11 by catalytic NaOMe in methanol in quantitative yield.
(-)-(6R,7S,8R,8aS)-6,7,8-Trihydroxyindolizidine ent-11 has been recently prepared by St -Denis and Chan, who claim $[\alpha]_{\mathrm{D}}^{20}-17.3 \times 10^{-1} \mathrm{deg} \mathrm{cm}{ }^{2} \mathrm{~g}^{-1}$ in methanol. ${ }^{6 q}$ This compound offered us the opportunity to test the feasibility of our synthetic procedure, as applied to a known target molecule. Indeed, starting from ent-1 and paralleling exactly the chemistry exploited for triol 11 , ent- 11 was prepared in $24 \%$ overall yield, via intermediates ent-2, ent-6, ent-4 and ent-10. The measured optical rotation in methanol was -19.1 at $22^{\circ} \mathrm{C}$, in accord with the reported value. Obviously, the spectroscopic properties of compound 11 and its enantiomer ent-11 were indistinguishable and matched well those reported for ent-11. ${ }^{6 q}$

Overall, concise asymmetric syntheses of the indolizidine 11 and its enantiomer ent-11 from tartaric acid-derived L- and Dthreoses 1 and ent-1 were established, involving only six (or eight) steps, in $22 \%$ and $26 \%$ yield, respectively.
$X$-Ray Study.-An X-ray analysis of compound 5 allowed its unambiguous structural determination establishing, as a consequence, the chirality of the related precursor 2 , the several intermediates, and the indolizidine 11. The absolute configuration of lactam 5 was determined as $4 R, 5 S, 6 S, 7 S$ based on the chirality of its precursor ( $2 R, 3 S$ )-4- $O$-benzyl-2,3- $O$-isopropyl-idene-L-threose 1 as in Scheme 2. An ORTEP view of the


Fig. 2 ORTEP drawing of compound 5 showing the atomicnumbering scheme. Thermal ellipsoids enclose $30 \%$ probability.
molecule in its correct absolute configuration is shown in Fig. 2.
The fractional atomic co-ordinates obtained from the crystallographic analysis and the subsequently derived interatomic distances and bond angles have been deposited as a supplementary publication.* In the solid-state structure the least-squares planes of the two rings form an angle of $88.7(2)^{\circ}$. The lactam ring assumes an envelope conformation with $q_{2}=$ $0.204(7) \AA, \varphi_{2}=-68(2)^{\circ},{ }^{7 a}$ while in the dioxolane ring is present a quasi-twist conformation with the diad axis passing through the $\mathrm{O}(4)$ oxygen atom $\left[q_{2}=0.298(5) \AA, \varphi_{2}=11(1)^{\circ}\right]$, maybe the most probable for rings with adjacent carbon atoms which are substituted. ${ }^{7 b}$ Both rings present significant displacements from their least-squares planes: a maximum deviation of $0.13 \AA$ for $C(2)$ in the lactam ring and of $0.19 \AA$ for $\mathrm{C}(6)$ in the dioxolane. The angle $\mathrm{C}(1)-\mathrm{N}(1)-\mathrm{C}(4)$, of $115.5(5)^{\circ}$, is similar to the values found in the literature for pyrrolidine rings with an $\mathrm{N}-\mathrm{H}$ bond. ${ }^{7 c}$ A certain charge delocalization can be observed in the $\mathrm{C}(4) \mathrm{N}(1) \mathrm{C}(1) \mathrm{O}(1)$ moiety of the lactam ring,

* Supplementary publication (see Instructions for Authors, in the January issue). Tables of atomic co-ordinates, bond lengths and bond angles have been deposited at the Cambridge Crystallographic Data Centre.


Fig. 3 Projection of the X-ray crystal structure of compound 5 along the $a$-axis
evidenced by both bond lengths and the torsion angle of $176.9(6)^{\circ} .^{7 c-e}$
The orientation of the OH group with respect to the pyrrolidine and dioxolane rings is shown by the following torsion angles: $\mathrm{N}(1) \mathrm{C}(4) \mathrm{C}(5) \mathrm{O}(2)=54.5(6)^{\circ}, \mathrm{C}(3) \mathrm{C}(4) \mathrm{C}(5)$ -$\mathrm{O}(2)=-61.1(6)^{\circ} ; \mathrm{O}(3) \mathrm{C}(6) \mathrm{C}(5) \mathrm{O}(2)=177(5)^{\circ}, \mathrm{C}(7) \mathrm{C}(6) \mathrm{C}(5)-$ $\mathrm{O}(2)=63.3(6)^{\circ}$. The distance $\mathrm{H}(6) \cdots \mathrm{H}(7)$ is $2.27(8) \AA$ and the torsion angle $\mathrm{H}(6) \mathrm{C}(4) \mathrm{C}(5) \mathrm{H}(7)=-50(5)^{\circ}$. corresponds to a synclinal conformation.

The distance $\mathrm{H}(7) \cdots \mathrm{H}(8)=1.94(8) \AA$ is extremely short, probably as a consequence of the excessive closeness of the $\mathrm{H}(8)$ hydrogen to the $O(2)$ oxygen atom, and the torsion angle of $-38(7)^{\circ}$ corresponds to an intermediate conformation between synperiplanar and synclinal. The $\mathrm{C}-\mathrm{O}$ distances in the dioxolane ring are similar to the values found in the literature. ${ }^{16,2 b, 7 b}$ The orientation of the tert-butyl ether group is evidenced by the following torsion angles: $\mathrm{O}(4) \mathrm{C}(7) \mathrm{C}(11) \mathrm{O}(5)=$ $-64.9(7), \mathrm{C}(6) \mathrm{C}(7) \mathrm{C}(11) \mathrm{O}(5)=178.7(5), \mathrm{C}(7) \mathrm{C}(11) \mathrm{O}(5) \mathrm{C}(12)$ $=-172.8(6)^{\circ}$

The packing is determined by a strong hydrogen bond $\mathrm{O}(2)-\mathrm{H}(8) \cdots \mathrm{O}(1)(\mathrm{I})=2.716(7) \AA(\mathrm{I}=1-x, 1 / 2+y, 1 / 2-$ $z$ ) between molecules related by a diad screw-axis. The helicoidal chains so formed stretching along the $y$-axis are joined among themselves by a weaker hydrogen bond $\mathrm{N}(1)-\mathrm{H}(1) \cdots \mathrm{O}(4)(\mathrm{II})=3.277(7) \AA(\mathrm{II}=1-x, y-1 / 2$, $3 / 2-z$ ) (Fig. 3).

## Experimental

M.p.s. were determined on a Büchi melting-point apparatus and are uncorrected. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra were performed with a Varian XL 300 spectrometer. $J$ Values are given in Hz and chemical-shift values in ppm referred to tetramethylsilane ( 0.0 ppm ), $\mathrm{DOH}(4.80 \mathrm{ppm}), \mathrm{CD}_{3} \mathrm{OD}$ ( 3.35 and 49.0 ppm ) and 1,4-dioxane ( 67.4 ppm ). $N$-tert-Butoxycarbonyl-2-(tert-butyldimethylsiloxy)pyrrole (TBSOP) was prepared from pyrrole according to a recently described protocol. ${ }^{2}$ 4-O-Benzyl-2,3-O-isopropylidene-L- and D-threose 1 and ent-1 were prepared from the corresponding diethyl tartrates according to literature. ${ }^{3}[\alpha]_{\mathrm{D}}$ Values were measured on a Perkin-Elmer 241 polarimeter, and are given in units of $10^{-1} \mathrm{deg} \mathrm{cm}^{2} \mathrm{~g}^{-1}$. Elemental analyses were performed by the Microanalytical Laboratory of the University of Sassari.

4-Amino-8-O-benzyl-N-(tert-butoxycarbonyl)-2,3,4-trideoxy-6,7-O-isopropylidene-L-galacto-oct-2-enonic Acid 1,4-Lactam 2.-TBSOP ( $3.35 \mathrm{~g}, 11.27 \mathrm{mmol}$ ) and L-threose derivative 1 $(2.32 \mathrm{~g}, 11.27 \mathrm{mmol})$ were dissolved in anhydrous $\mathrm{Et}_{2} \mathrm{O}\left(50 \mathrm{~cm}^{3}\right)$ under argon and the mixture was cooled to $-80^{\circ} \mathrm{C}$. A 1 mol $\mathrm{dm}^{-3}$ solution of $\mathrm{SnCl}_{4}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(16.90 \mathrm{mmol}, 16.90 \mathrm{~cm}^{3}\right)$ was added at room temperature, via a cannula, during 10 min and the solution was stirred for 5 h . The reaction was quenched at this temperature by addition of an excess of saturated aq. $\mathrm{NaHCO}_{3}$. Then the mixture was warmed to room temperature and extracted with $\mathrm{Et}_{2} \mathrm{O}\left(3 \times 15 \mathrm{~cm}^{3}\right)$. The organic layer was dried over $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure to furnish the crude lactam, which was purified by flash chromatography over silica gel and eluted with 65:35 EtOAchexane to afford lactam $2(3.42 \mathrm{~g}, 80 \%)$ as a solid, m.p. $96-98^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}^{20}+134.66\left(c 0.88\right.$ in $\left.\mathrm{CHCl}_{3}\right)$ (Found: C, $63.65 ; \mathrm{H}, 7.1 ; \mathrm{N}$, 3.15. $\mathrm{C}_{23} \mathrm{H}_{31} \mathrm{NO}_{7}$ requires C, $\left.63.73 ; \mathrm{H}, 7.21 ; \mathrm{N}, 3.23 \%\right)$; $\delta_{\mathrm{H}}(300$ $\mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) $1.28(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 1.30(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 1.53(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{Bu}^{t}\right), 3.52(1 \mathrm{H}$, dd, $J 8.8$ and $7.5,6-\mathrm{H}), 3.53(1 \mathrm{H}$, dd, $J 9.3$ and $6.9,8-\mathrm{H}^{\mathrm{b}}$ ), $3.73\left(1 \mathrm{H}, \mathrm{dd}, J 9.3\right.$ and $\left.4.5,8-\mathrm{H}^{\mathrm{a}}\right), 4.07(1 \mathrm{H}, \mathrm{dd}, J 6.9$ and $4.5,7-\mathrm{H}), 4.15(1 \mathrm{H}$, ddd, $J 9.0,5.1$ and $3.3,5-\mathrm{H}), 4.30$ $(1 \mathrm{H}, \mathrm{d}, J 3.0, \mathrm{OH}), 4.58\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.76(1 \mathrm{H}, \mathrm{dt}, J 4.8$ and $1.8,4-\mathrm{H}), 6.12(1 \mathrm{H}, \mathrm{dd}, J 6.0$ and $1.8,2-\mathrm{H})$ and $7.31(6 \mathrm{H}, \mathrm{m}, P h$, $3-\mathrm{H}) ; \delta_{\mathrm{C}}\left(75.4 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 26.52,28.05,65.65,70.54,72.38$,
$73.83,78.93,79.56,82.99,109.86,127.65,128.09,136.78,150.31$ and 169.06 .

4-Amino-N-(tert-butoxycarbonyl)-2,3,4-trideoxy-6,7-O-iso-propylidene-L-galacto-octonic Acid 1,4-Lactam 3.-To a solution of unsaturated lactam $2(1 \mathrm{~g}, 2.3 \mathrm{mmol})$ in THF $\left(40 \mathrm{~cm}^{3}\right)$ were added $\mathrm{Pd}-\mathrm{C}(100 \mathrm{mg})$ and $\mathrm{AcONa}(50 \mathrm{mg})$. The mixture was stirred for 48 h at room temperature, then was filtered and the resulting solution was evaporated under reduced pressure. The crude product was purified by flash chromatography over silica gel and eluted with EtOAc to afford compound $\mathbf{3}(755 \mathrm{mg}$, $95 \%$ ) as an oil; $[\alpha]_{\mathrm{D}}^{20}+45.83$ (c 0.24 in $\mathrm{CHCl}_{3}$ ) (Found: C, 55.4; $\mathrm{H}, 7.8 ; \mathrm{N}, 4.0 . \mathrm{C}_{16} \mathrm{H}_{27} \mathrm{NO}_{7}$ requires $\mathrm{C}, 55.62 ; \mathrm{H}, 7.88 ; \mathrm{N}, 4.06 \%$ ); $\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.39(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 1.41(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 1.52$ ( $9 \mathrm{H}, \mathrm{s}, \mathrm{Bu}^{t}$ ), 2.28-2.07 ( $2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}_{2}$ ), 2.37 ( $1 \mathrm{H}, \mathrm{ddd}, J 17.7,9.0$ and $1.5,2-\mathrm{H}^{\mathrm{b}}$ ), $2.66\left(1 \mathrm{H}\right.$, ddd, $J 16.2,12.0$ and $9.0,2-\mathrm{H}^{\mathrm{a}}$ ), $3.81-$ $3.70\left(3 \mathrm{H}, \mathrm{m}, 5-\mathrm{and} 6-\mathrm{H}, 8-\mathrm{H}^{\mathrm{b}}\right), 3.86(1 \mathrm{H}, \mathrm{dd}, J 11.7$ and 3.9 , $\left.8-\mathrm{H}^{\mathrm{a}}\right), 4.08(1 \mathrm{H}, \mathrm{td}, J 6.3$ and $3.9,7-\mathrm{H})$ and $4.37(1 \mathrm{H}, \mathrm{ddd}, J 8.1$, 5.7 and $1.8,4-\mathrm{H}) ; \delta_{\mathrm{C}}\left(75.4 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right.$ ) 20.87, 26.68. 26.82, 27.97, 31.81, 61.20, 63.07, 74.29, 79.34, 80.93, 83.62, 109.54, 151.76 and 174.90 .

4-Amino-8-O-(tert-butyl)-2,3,4-trideoxy-6,7-O-isopropyli-dene-L-galacto-octonic Acid 1,4-Lactam 5.-To a solution of compound 3 ( $500 \mathrm{mg}, 1.45 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(30 \mathrm{~cm}^{3}\right)$ under argon at $\mathrm{O}^{\circ} \mathrm{C}$ was added TMSOTf ( $644.5 \mathrm{mg}, 2.9 \mathrm{mmol}$ ). The mixture was stirred at this temperature for 6 h , then saturated aq. $\mathrm{NaHCO}_{3}$ was added. The aqueous layer was extracted with EtOAc and the organic phase, dried over $\mathrm{MgSO}_{4}$, was evaporated under reduced pressure. The crude product was purified by flash chromatography on silica gel and eluted with 8:2 EtOAc-MeOH to afford compound $5(411 \mathrm{mg}, 94 \%)$ as crystals, m.p. $170-171{ }^{\circ} \mathrm{C} ;[\alpha]_{D}^{24}-11.1$ (c 0.20 in $\mathrm{CHCl}_{3}$ ) (Found: C, 59.7; H, 9.1; N, 4.6. $\mathrm{C}_{15} \mathrm{H}_{27} \mathrm{NO}_{5}$ requires C, 59.76; $\mathrm{H}, 9.03 ; \mathrm{N}, 4.65 \%) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.23\left(9 \mathrm{H}, \mathrm{s}, \mathrm{Bu}^{t}\right), 1.37$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Me}$ ), $1.38(3 \mathrm{H}, \mathrm{s} . \mathrm{Me}), 2.10\left(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}^{\mathrm{b}}\right), 2.26(2 \mathrm{H}, \mathrm{m}$, $\left.3-\mathrm{H}_{2}\right), 2.44\left(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}^{\mathrm{a}}\right), 3.34\left(1 \mathrm{H}, \mathrm{t}, \mathrm{J} 8.7,8-\mathrm{H}^{\mathrm{b}}\right), 3.46(1 \mathrm{H}$, $\mathrm{m}, 5-\mathrm{H}), 3.63(1 \mathrm{H}, \mathrm{t}, J 7.8,6-\mathrm{H}), 3.77(1 \mathrm{H}, \mathrm{dd}, J 8.4$ and $3.9,8-$ $\left.\mathrm{H}^{\mathrm{a}}\right), 3.84(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 3.49(1 \mathrm{H}, \mathrm{dq}, J 9.0,7.5$ and $3.9,7-\mathrm{H})$, $4.21(1 \mathrm{H}, \mathrm{d}, J 1.8, \mathrm{OH})$ and $6.02(1 \mathrm{H}, \mathrm{s}, \mathrm{NH})$.

## 4-Amino-8-O-benzyl-2,3,4-trideoxy-6,7-O-isopropylidene-

 L-galacto-oct-2-enonic Acid 1,4-Lactam 6.-To a solution of compound $2(3.0 \mathrm{~g}, 6.92 \mathrm{mmol})$ in anhydrous $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(130 \mathrm{~cm}^{3}\right)$ were added, under argon, at $0^{\circ} \mathrm{C}, \operatorname{PhSH}(1.14 \mathrm{~g}, 10.38 \mathrm{mmol})$ and TMSOTf ( $2.307 \mathrm{~g}, 10.38 \mathrm{mmol}$ ). The mixture was stirred for 1 h at this temperature, then was quenched by addition of saturated aq. $\mathrm{NaHCO}_{3}$. The aqueous layer was extracted with EtOAc and the organic phase was evaporated under reduced pressure. The crude product was then purified by flash chromatography over silica gel and eluted with 95:5 EtOAcMeOH to afford title compound $6(1.73 \mathrm{~g}, 75 \%)$ as an oil; $[\alpha]_{\mathrm{D}}^{25}$ +23.49 (c 3.1 in EtOAc) (Found: C, 64.8; H, 6.8; N, 4.35. $\mathrm{C}_{18} \mathrm{H}_{23} \mathrm{NO}_{5}$ requires $\mathrm{C}, 64.85 ; \mathrm{H}, 6.95 ; \mathrm{N}, 4.20 \%$ ); $\delta_{\mathrm{H}}(300 \mathrm{MHz}$; $\mathrm{CDCl}_{3}$ ) $1.36(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 1.41(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 3.48(1 \mathrm{H}, \mathrm{td}, J 6.6$ and $6.3,5-\mathrm{H}), 3.64\left(1 \mathrm{H}\right.$, dd, $J 9.0$ and $\left.6.6,8-\mathrm{H}^{\mathrm{b}}\right), 3.70(1 \mathrm{H}, \mathrm{dd}, J$ 6.6 and $4.5,6-\mathrm{H}), 3.83\left(1 \mathrm{H}, \mathrm{t}, J 9,8-\mathrm{H}^{\mathrm{a}}\right), 4.18(1 \mathrm{H}, \mathrm{m}, 7-\mathrm{H})$, $4.29(1 \mathrm{H}, \mathrm{d}, \mathrm{J} 6.6,4-\mathrm{H}), 4.58\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.77(1 \mathrm{H}, \mathrm{d}, J 6.3$, $\mathrm{OH}), 6.05(1 \mathrm{H}, \mathrm{d}, J 5.7,2-\mathrm{H}), 7.17(1 \mathrm{H}, \mathrm{d}, J 5.7,3-\mathrm{H}), 7.29(5 \mathrm{H}$, $\mathrm{m}, \mathrm{Ph})$ and $7.90(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}) ; \delta_{\mathrm{c}}\left(75.4 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 26.83$, $26.89,64.13,70.87,73.41,73.84,78.95,79.41,109.79,127.26$, 127.62, 128.22, 137.51 and 175.06 .
## 4-Amino-2,3,4-trideoxy-6,7-O-isopropylidene-L-galacto-

 octonic Acid 1,4-Lactam 4.-To a solution of compound 6 $(1.8 \mathrm{~g}, 5.4 \mathrm{mmol})$ in THF ( $80 \mathrm{~cm}^{3}$ ) were added $\mathrm{Pd}-\mathrm{C}(200 \mathrm{mg})$ and $\mathrm{NaOAc}(100 \mathrm{mg})$. The mixture was stirred for 48 h under$\mathrm{H}_{2}$, then the solution was filtered, and evaporated under reduced pressure. The crude product was purified by flash chromatography over silica gel and eluted with $8: 2$ EtOAcMeOH to afford saturated lactam $4(1.26 \mathrm{~g}, 95 \%)$ as an oil; $[\alpha]_{\mathrm{D}}^{21}$ -24.28 (c 0.7 in MeOH) (Found: C, 53.6; H, 7.8; N, 5.6. $\mathrm{C}_{11} \mathrm{H}_{19} \mathrm{NO}_{5}$ requires $\left.\mathrm{C}, 53.87 ; \mathrm{H}, 7.81 ; \mathrm{N}, 5.71 \%\right) ; \delta_{\mathrm{H}}(300 \mathrm{MHz}$; $\mathrm{CDCl}_{3}$ ) $1.35(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 1.37(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 2.43-2.02(4 \mathrm{H}, \mathrm{m}, 2-$ and $\left.3-\mathrm{H}_{2}\right), 3.42(1 \mathrm{H}, \mathrm{td}, J 8.4$ and $3.6,5-\mathrm{H}), 3.66(3 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}$, $8-\mathrm{H}_{2}$ ), $3.82(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 3.96(1 \mathrm{H}, \mathrm{m}, 7-\mathrm{H}), 5.31(2 \mathrm{H}, \mathrm{br}, \mathrm{OH})$ and $6.80(1 \mathrm{H}, \mathrm{s}, \mathrm{NH}) ; \delta_{\mathrm{C}}\left(75.4 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 23.41,26.81,30.54$, $56.90,62.57,75.19,78.17,80.70,109.23$ and 180.25 .

4-Amino-2,3,4-trideoxy-6,7-O-isopropylidene-8-O-methylsul-fonyl-L-galacto-octonic Acid 1,4-Lactam 7.-To a solution of compound 6 ( $500 \mathrm{mg}, 2.04 \mathrm{mmol}$ ) in anhydrous $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(10 \mathrm{~cm}^{3}\right)$ under argon at room temperature were sequentially added pyridine ( $483 \mathrm{mg}, 6.12 \mathrm{mmol}$ ), $\mathrm{MsCl}(257 \mathrm{mg}, 2.24 \mathrm{mmol})$ and DMAP ( $25 \mathrm{mg}, 0.20 \mathrm{mmol}$ ). The mixture was stirred for 7 h , then was quenched with water and evaporated under reduced pressure. The crude product was purified by flash chromatography over silica gel and eluted with $8: 2 \mathrm{AcOEt}-\mathrm{MeOH}$ to afford the mesate $7(488 \mathrm{mg}, 74 \%$ ) as an oil (Found: C, 44.3; H, 6.55; $\mathrm{N}, 4.65 . \mathrm{C}_{12} \mathrm{H}_{21} \mathrm{NO}_{7} \mathrm{~S}$ requires $\mathrm{C}, 44.57 ; \mathrm{H}, 6.55 ; \mathrm{N}, 4.33 \%$ ); $\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.40(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 1.42(3 \mathrm{H}, \mathrm{s}, \mathrm{Me})$, $2.14-2.06\left(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}^{\mathrm{b}}\right), 2.40-2.20\left(3 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}_{2}, 3-\mathrm{H}^{\mathrm{a}}\right), 3.09$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{SMe}$ ), 3.47 ( 1 H , ddd, $J 8.4,8.0$ and $5.0 .5-\mathrm{H}$ ), 3.81 ( 1 H , dd, $J 8.4,6-\mathrm{H}), 3.87(1 \mathrm{H}, \mathrm{q}, J 5.1,4-\mathrm{H}), 4.38-4.22(2 \mathrm{H}, \mathrm{m}, 7-\mathrm{H}$, $\left.8-\mathrm{H}^{\mathrm{b}}\right), 4.54\left(1 \mathrm{H}, \mathrm{dd}, J 11.1\right.$ and $\left.1.8,8-\mathrm{H}^{\mathrm{a}}\right), 4.72(1 \mathrm{H}, \mathrm{d}, J 7.9$, $\mathrm{OH})$ and $7.19(1 \mathrm{H}, \mathrm{s}, \mathrm{NH})$.
(2R)-2-( $\alpha$-L-threo-Pentofuranosyl)pyrrolidine 8.-To a solution of lactam $7(500 \mathrm{mg}, 1.55 \mathrm{mmol})$ in anhydrous THF ( 20 $\mathrm{cm}^{3}$ ), under argon at room temperature, was added a 10 mol $\mathrm{dm}^{-3}$ solution of $\mathrm{BH}_{3} \cdot \mathrm{DMS}\left(7.75 \mathrm{~cm}^{3}, 77.5 \mathrm{mmol}\right)$. The mixture was stirred for 15 h , then quenched with MeOH and evaporated under reduced pressure. The crude product was treated with $\mathrm{CF}_{3} \mathrm{CO}_{2} \mathrm{H}$-water $4: 1\left(12 \mathrm{~cm}^{3}\right)$, then was purified by passage through a DOWEX ( $\mathrm{OH}^{-}$form) column eluted with water, to afford compound 8 ( $120 \mathrm{mg}, 45 \%$ ) as an oil (Found: C, $55.4 ; \mathrm{H}$, $8.4 ; \mathrm{N}, 7.85 \mathrm{C}_{8} \mathrm{H}_{15} \mathrm{NO}_{3}$ requires $\mathrm{C}, 55.47 ; \mathrm{H}, 8.73 ; \mathrm{N}, 8.09 \%$ ); $\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{D}_{2} \mathrm{O}\right) 1.82-1.32\left(4 \mathrm{H}, \mathrm{m}, 2\right.$ - and $\left.3-\mathrm{H}_{2}\right), 2.62(2 \mathrm{H}$, $\left.\mathrm{m}, 1-\mathrm{H}_{2}\right), 3.08(1 \mathrm{H}, \mathrm{q}, J 7.5,4-\mathrm{H}), 3.47(1 \mathrm{H}, \mathrm{dd}, J 7.0$ and 3.0 , $5-\mathrm{H}), 3.63\left(1 \mathrm{H}\right.$, ddd, $J 10.2,1.5$ and $\left.0.9,8-\mathrm{H}^{\mathrm{b}}\right), 3.77(1 \mathrm{H}, \mathrm{dd}, J$ 10.2 and $3.6,8-\mathrm{H}^{2}$ ), $3.82(1 \mathrm{H}$, ddd, $J 3.0,1.5$ and $0.96-\mathrm{H})$ and $3.94(1 \mathrm{H}, \mathrm{dt}, J 3.9$ and $1.5,7-\mathrm{H}) ; \delta_{\mathrm{C}}\left(75.4 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}\right) 27.56$, $29.08,47.72,61.01,75.00,78.63,81.18$ and 90.48 .

1,2,3,4-Tetradeoxy-1,4-imino-6-O-isopropyl-L-galacto-octitol 9.-To a solution of compound $4(1.0 \mathrm{~g}, 4.01 \mathrm{mmol})$ in anhydrous THF ( $60 \mathrm{~cm}^{3}$ ) under argon was added a 10 mol $\mathrm{dm}^{-3}$ solution of $\mathrm{BH}_{3} \cdot$ DMS complex ( $28.6 \mathrm{~cm}^{3}, 285.6 \mathrm{mmol}$ ) at room temperature. The mixture was stirred at this temperature for 24 h , then MeOH was added and the solution was evaporated under reduced pressure. The crude product was purified by flash chromatography over silica gel and eluted with 9:1 EtOAc-MeOH to afford an oil ( $698 \mathrm{mg}, 74 \%$ ). To this material was added $2 \mathrm{~mol} \mathrm{dm}{ }^{-3} \mathrm{HCl}\left(10 \mathrm{~cm}^{3}\right)$, then the solution was stirred for 30 min and evaporated under reduced pressure, and the crude product was purified by passage through a DOWEX ( $\mathrm{OH}^{-}$form) column to afford compound $9(676 \mathrm{mg}$, $96 \%$ ) as an oil; $[\alpha]_{\mathrm{D}}^{19}+1.54$ (c 1.3 in MeOH ) (Found: C, 56.6; $\mathrm{H}, 9.9 ; \mathrm{N}, 6.1 \mathrm{C}_{11} \mathrm{H}_{23} \mathrm{NO}_{4}$ requires C, $56.63 ; \mathrm{H}, 9.94 ; \mathrm{N}, 6.0 \%$ ); $\left.\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{D}_{2} \mathrm{O}\right) 1.21(6 \mathrm{H}, \mathrm{t}, \mathrm{CHMe})_{2}\right), 1.63\left(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}^{\mathrm{b}}\right)$, $1.83\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}_{2}\right), 1.92\left(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}^{2}\right), 2.90\left(1 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}^{\mathrm{b}}\right)$, $3.04\left(1 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}^{\mathrm{a}}\right), 3.30(1 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 3.52(1 \mathrm{H}, \mathrm{dd}, J 5.4$ and $2.7,6-\mathrm{H}), 3.66\left(2 \mathrm{H}, \mathrm{m}, 8-\mathrm{H}_{2}\right), 3.72(1 \mathrm{H}, \mathrm{t}, J 5.4,5-\mathrm{H})$ and 3.88 ( $2 \mathrm{H}, \mathrm{m}, 7-\mathrm{H}, \mathrm{CHMe} 2$ ); $\delta_{\mathrm{C}}\left(75.4 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}\right.$ ) 22.82, 23.20, $26.25,29.53,47.23,61.03,63.91,73.08,73.54,74.73$ and 78.01 .
( $6 \mathrm{~S}, 7 \mathrm{R}, 8 \mathrm{~S}, 8 \mathrm{aR}$ )-6,8-Dihydroxy-7-isopropyloxyindolizidine 10.-Under rigorously anhydrous conditions in a vessel shielded from light, to a solution of compound $9(260 \mathrm{mg}, 1.11$ mmol ) in pyridine ( $6 \mathrm{~cm}^{3}$ ) under argon at room temperature were added $\mathrm{PPh}_{3}(716 \mathrm{mg}, 2.72 \mathrm{mmol}), \mathrm{CCl}_{4}\left(139 \mathrm{~mm}^{3}, 1.36\right.$ $\mathrm{mmol})$ and, after $3 \mathrm{~h}, \mathrm{Et}_{3} \mathrm{~N}\left(379 \mathrm{~mm}^{3}, 2.72 \mathrm{mmol}\right)$. The mixture was stirred overnight then was quenched with $\mathrm{MeOH}\left(5 \mathrm{~cm}^{3}\right)$ and evaporated under reduced pressure. The crude product was purified by passage first through a DOWEX 50 W X $8\left(\mathrm{H}^{+}\right.$ form) column eluted successively with MeOH , water and aq. $\mathrm{NH}_{4} \mathrm{OH}$, then by flash chromatography over silica gel and elution with $2: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ to afford compound $\mathbf{1 0}$ (188 $\mathrm{mg}, 79 \%$ ) as an oil; $[\alpha]_{\mathrm{D}}^{19}+21.1$ ( $c 0.80$ in MeOH) (Found: C, 61.15; $\mathrm{H}, 9.65 ; \mathrm{N}, 6.3 . \mathrm{C}_{11} \mathrm{H}_{21} \mathrm{NO}_{3}$ requires $\mathrm{C}, 61.37 ; \mathrm{H}, 9.83 ; \mathrm{N}$, $6.51 \%) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{D}_{2} \mathrm{O}\right) 1.15(6 \mathrm{H}, \mathrm{m}), 2.04-1.75(4 \mathrm{H}, \mathrm{m})$, $2.14(2 \mathrm{H}, \mathrm{m}), 2.87(1 \mathrm{H}, \mathrm{m}), 3.10(1 \mathrm{H}, \mathrm{dd}, J 10.8$ and 10.4$), 3.28$ ( $1 \mathrm{H}, \mathrm{dd}, J 9.9$ and 3.6 ), $3.76(3 \mathrm{H}, \mathrm{m})$ and $4.03(1 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}(75.4$ $\mathrm{MHz} ; \mathrm{D}_{2} \mathrm{O}$ ) 22.69, 23.08, 24.18, 25.47, 54.51, 57.18, 67.08, 67.40, 68.20, 72.39 and 82.74.
(6S,7R,8S,8aR)-6,7,8-Trihydroxyindolizidine 11.-To a stirred solution of the ether $10(150 \mathrm{mg}, 0.70 \mathrm{mmol})$ in anhydrous $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(8 \mathrm{~cm}^{3}\right)$ was added $\mathrm{BBr}_{3}\left(0.65 \mathrm{~cm}^{3}, 7.0\right.$ mmol at $-78^{\circ} \mathrm{C}$ and the mixture was treated for 20 h at room temperature. The reaction was quenched with saturated aq. $\mathrm{NH}_{4} \mathrm{OH}$ and the resulting slurry was concentrated under reduced pressure. The residue was purified by flash chromatography over silica gel and eluted with $\mathrm{EtOH}-\mathrm{CH}_{2} \mathrm{Cl}_{2}-$ $\mathrm{NH}_{4} \mathrm{OH}$ 10:5:1 to give the indolizidine $11(81 \mathrm{mg}, 67 \%)$ as a glass; $[\alpha]_{\mathrm{D}}^{20}+17.5$ (c 0.30 in MeOH) (Found: C, 55.35 ; H, 8.6; $\mathrm{N}, 8.2 . \mathrm{C}_{8} \mathrm{H}_{15} \mathrm{NO}_{3}$ requires $\mathrm{C}, 55.47 ; \mathrm{H}, 8.73 ; \mathrm{N}, 8.09 \%$ ); $\delta_{\mathrm{H}}(300$ $\left.\mathrm{MHz} ; \mathrm{D}_{2} \mathrm{O}\right) 1.77(3 \mathrm{H}, \mathrm{m}), 2.0(1 \mathrm{H}, \mathrm{m}), 2.20(2 \mathrm{H}, \mathrm{m}), 2.93(1 \mathrm{H}$, brt, $J 6.0$ ), $3.17(1 \mathrm{H}, \mathrm{dd}, J 10.5$ and 5.1$), 3.38(1 \mathrm{H}, \mathrm{m}), 3.47(1 \mathrm{H}$, dd, $J 9.9$ and 3.9), $3.82(1 \mathrm{H}, \mathrm{td}, J 10.2$ and 5.1$)$ and $3.95(1 \mathrm{H}, \mathrm{dd}$, $J 9.9$ and 3.9 ); $\delta_{\mathrm{C}}\left(75.4 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}\right.$ ) 23.00, 25.31, 54.59, 57.78, 67.82, 69.35, 69.85 and 77.99.
(6S,7R,8S,8aR)-6.8-Diacetoxy-7-isopropoxyindolizidine 12. -To a stirred solution of the isopropyl ether $\mathbf{1 0}(54 \mathrm{mg}, 0.31$ $\mathrm{mmol})$ in dry pyridine $\left(1.0 \mathrm{~cm}^{3}\right)$ were added $\mathrm{Ac}_{2} \mathrm{O}\left(20 \mathrm{~cm}^{3}\right)$ and DMAP ( 20 mg ) at room temperature. After 5 h the solvent was removed and the residue was purified by flash chromatography over silica gel and eluted with 7:3 EtOAc-hexane to afford diacetate 12 ( $92 \mathrm{mg}, 100 \%$ ) as a glass (Found: C, $60.15 ; \mathrm{H}$, $8.2 ; \mathrm{N}, 4.7 \mathrm{C}_{15} \mathrm{H}_{25} \mathrm{NO}_{5}$ requires $\mathrm{C}, 60.18 ; \mathrm{H}, 8.42 ; \mathrm{N}, 4.68 \%$ ); $\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.10$ and $\left.1.08(2 \times 3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6.0, \mathrm{CHMe})_{2}\right)$, $1.1-2.2(7 \mathrm{H}, \mathrm{m}), 2.17$ and $2.04(2 \times 3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 3.10(1 \mathrm{H}, \mathrm{td}, J$ 9.9 and 1.8 ), $3.38(1 \mathrm{H}, \mathrm{dd}, J 10.2$ and $3.9,7-\mathrm{H}), 3.41(1 \mathrm{H}, \mathrm{dd}, J$ 10.2 and $5.1,5-\mathrm{H}^{\text {eq }}$ ), $3.67\left(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{Me}_{2}\right), 5.10(1 \mathrm{H}, \mathrm{td}, J 10.2$ and $5.1,6-\mathrm{H})$ and $5.45(1 \mathrm{H}$, dd, $J 3.6$ and $1.8,8-\mathrm{H}) ; \delta_{\mathrm{C}}(74.5$ $\mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) 20.93, 21.00, 21.53, 22.91, 24.70, 29.63, 53.23, $53.59,64.44,67.61,69.71,70.94,77.49,170.04$ and 171.20.
(6S,7R,8S,8aR)-6,8-Diacetoxy-7-hydroxyindolizidine 13.To a stirred solution of the ether $12(30 \mathrm{mg}, 0.1 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $\left(1 \mathrm{~cm}^{3}\right)$ at $-78{ }^{\circ} \mathrm{C}$ was added $\mathrm{BBr}_{3}\left(0.1 \mathrm{~cm}^{3}\right)$ and the mixture was allowed to react for 20 h at room temperature. The reaction was quenched with saturated aq. $\mathrm{NH}_{4} \mathrm{OH}$ and the resulting mixture was evaporated under reduced pressure. The residue was purified by flash chromatography over silica gel and eluted with 9: 1 AcOEt-MeOH to afford compound $13(24 \mathrm{mg}, 95 \%)$ as a glass (Found: C, $55.9 ; \mathbf{H}, 7.4 ; \mathrm{N}, 5.3 \mathrm{C}_{12} \mathrm{H}_{19} \mathrm{NO}_{5}$ requires C , $56.02,7.44 ; \mathrm{N}, 5.44 \%) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) 2.05-1.60(4 \mathrm{H}$, $\mathrm{m}), 2.10$ and $1.96(2 \times 3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 2.31(1 \mathrm{H}, \mathrm{m}), 3.07(2 \mathrm{H}, \mathrm{m})$, $3.26(1 \mathrm{H}, \mathrm{dd}, J 11.1$ and 5.0$), 3.51(1 \mathrm{H}, \mathrm{dd}, J 10.0$ and 3.2 ), 3.86 $(1 \mathrm{H}, \mathrm{m}), 5.31(1 \mathrm{H}, \mathrm{m})$ and $5.35(1 \mathrm{H}, \mathrm{m})$. To a solution of acetate $13(20 \mathrm{mg} .0 .08 \mathrm{mmol})$ in methanol $\left(0.5 \mathrm{~cm}^{3}\right)$ were added a few drops of methanolic NaOMe at room temperature and
the mixture was allowed to react for 1 h . Removal of the solvent and DOWEX $\left(\mathrm{OH}^{-}\right)$column purification, elution with water, afforded the pure indolizidine $11(\sim 100 \%)$ which was identical in all respects $[\alpha]_{\mathrm{D}},{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR with the above compound.

4-(Amino-8-O-benzyl-N-(tert-butoxycarbonyl)-2,3,4-trideoxy-6,7-O-isopropylidene-D-galacto-oct-2-enonic Acid 1,4Lactam ent-2.-The title compound was prepared by starting with ent $-1(2.0 \mathrm{~g}, 9.7 \mathrm{mmol})$ following the procedure described for its enantiomer 2. Yield $3.27 \mathrm{~g}(78 \%) ;[\alpha]_{\mathrm{D}}^{20}-129.3$ (c 1.2 in $\mathrm{CHCl}_{3}$ )(Found: C, $63.6 ; \mathrm{H}, 7.15 ; \mathrm{N}, 3.15 . \mathrm{C}_{23} \mathrm{H}_{31} \mathrm{NO}_{7}$ requires C, 63.73; $\mathrm{H}, 7.21 ; \mathrm{N}, 3.23 \%) ;{ }^{1} \mathrm{H}$ and ${ }^{3} \mathrm{C}$ NMR, see compound 2.

4-Amino-8-O-benzyl-2,3,4-trideoxy-6,7-O-isopropylidene-D-galacto-oct-2-enonic Acid 1,4-Lactam ent-6.-The title compound was prepared by starting with ent-2 $(2.0 \mathrm{~g}, 4.6 \mathrm{mmol})$ and following the procedure described for its enantiomer 6. Yield $1.16 \mathrm{~g}(76 \%) ;[\alpha]_{\mathrm{D}}^{20}-22.5$ (c 2.0 in EtOAc) (Found: C, 64.9; H, 6.8; N, 4.1. $\mathrm{C}_{18} \mathrm{H}_{23} \mathrm{NO}_{5}$ requires $\mathrm{C}, 64.85 ; \mathrm{H}, 6.95 ; \mathrm{N}$, $4.20 \%) ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see compound 6.

4-Amino-2,3,4-trideoxy-6,7-O-isopropylidene-D-galactooctonic Acid 1,4-Lactam ent-4.-The title compound was prepared by starting with ent $-6(1.5 \mathrm{~g}, 4.5 \mathrm{mmol})$ and following the procedure described for its enantiomer 4. Yield $1.0 \mathrm{~g}(95 \%)$; $[\alpha]_{\mathrm{D}}^{20}+25.1$ (c $1.1 \mathrm{in} \mathrm{MeOH)} \mathrm{(Found:} \mathrm{C} ,54.0 ; \mathbf{H}, 7.8 ; \mathrm{N}$, 5.65. $\mathrm{C}_{11} \mathrm{H}_{19} \mathrm{NO}_{5}$ requires C, $53.87 ; \mathrm{H}, 7.81 ; \mathrm{N}, 5.71 \%$ ); ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see compound 4.

1,2,3,4-Tetradeoxy-1-4-imino-6-O-isopropyl-D-galacto-octitol ent-9.-The title compound was prepared by starting with ent-4 ( $800 \mathrm{mg}, 3.26 \mathrm{mmol}$ ) and following the procedure described for its enantiomer 9. Yield $676 \mathrm{mg}(76 \%)$; $[\alpha]_{\mathrm{D}}^{20}-1.7$ (c 1.0 in MeOH ) (Found: C, 55.7; $\mathrm{H}, 9.8 ; \mathrm{N}, 5.8 . \mathrm{C}_{11} \mathrm{H}_{23} \mathrm{NO}_{4}$ requires C, $55.63 ; \mathrm{H}, 9.94 ; \mathrm{N}, 6.0 \%) ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see compound 9 .
(6R,7S,8R,8aS)-6,8-Dihydroxy-7-isopropoxyindolizidine ent-10.-The title compound was prepared by starting with ent-9 ( $500 \mathrm{mg}, 2.14 \mathrm{mmol}$ ) and following the procedure described for its enantiomer 10. Yield $368 \mathrm{mg}(80 \%)$; $[\alpha]_{\mathrm{D}}^{20}-20.3$ (c 1.1 in MeOH ) (Found: $\mathrm{C}, 61.2 ; \mathrm{H}, 9.8$; N, 6.5. $\mathrm{C}_{11} \mathrm{H}_{21} \mathrm{NO}_{3}$ requires C , $61.37 ; \mathrm{H}, 9.83 ; \mathrm{N}, 6.51 \%$ ); ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see compound 10.
(6R,7S,8R,8aS)-6,7,8-Trihydroxyindolizidine ent-11.-The title compound was prepared by starting with ent-10 $(200 \mathrm{mg}$, 0.93 mmol ) and following the procedure described for its enantiomer 11. Yield $112 \mathrm{mg}(70 \%) ;[\alpha]_{\mathrm{D}}^{20}-19.1$ (c 0.1 in $\mathrm{MeOH})\left\{\right.$ lit., ${ }^{6 q}[\alpha]_{\mathrm{D}}^{20}-17.3$ (c 0.85 in MeOH ) $\}$ (Found: C, $55.4 ; \mathrm{H}, 8.7 ; \mathrm{N}, 8.1 . \mathrm{C}_{8} \mathrm{H}_{15} \mathrm{NO}_{3}$ requires $\mathrm{C}, 55.47 ; \mathrm{H}, 8.73 ; \mathrm{N}$, $8.09 \%) ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, see compound 11.

X-Ray Crystal Structure Determination of Compound 5.Crystal data. $\mathrm{C}_{15} \mathrm{H}_{27} \mathrm{NO}_{5}, M=301.4$, prisms, orthorhombic, space group $P 2_{1} 2_{1} 2_{1}$ (No. 19), $a=16.663(5), b=10.488(4)$, $c=9.869(3) \AA, V=1725(1) \AA^{3}, Z=4, F(000)=656, D_{\mathrm{c}}=$ $1.161 \mathrm{~g} \mathrm{~cm}^{-3}, \mu(\mathrm{Cu}-\mathrm{K} \alpha)=6.752 \mathrm{~cm}^{-1}$.

Data collection. The intensity data were collected on a Siemens AED diffractometer over a 3-70 $\theta$ range; $h, 0-20 ; k$, $0-12 ; l, 0-12$, by using the $\mathrm{Cu}-\mathrm{K} \alpha$ radiation $(\lambda=1.54178)$ and $\theta-2 \theta$ scanning. Of the 1911 unique data measured, 1437 had $I>2 \sigma(I)$ and were used in the subsequent structural solution and refinement. The data were corrected for Lorentz and polarization effects, but not for absorption. The crystal dimensions were $0.47 \times 0.47 \times 0.23 \mathrm{~mm}$.

Structure solution. The structural determination was carried out by direct methods using the SHELX- $86{ }^{8 a}$ program. The structure was then refined by full-matrix least-squares methods
(SHELX-76) ${ }^{8 b}$ using anisotropic temperature factors for all the non-hydrogen atoms. Hydrogen atoms were located with a difference Fourier map with the exception of methyl and methylene hydrogens, which were calculated at idealized positions ( $d_{\mathrm{C}-\mathrm{H}} 1.008 \AA$ ). All H -atoms were included in the refinement. The weighting scheme adopted was $w=0.5792 /$ $\sigma^{2}\left(F_{\mathrm{o}}\right)+0.01395\left(F_{\mathrm{o}}\right)^{2}$. At convergence, the discrepancy indices $R$ and $R_{w}$ were 0.067 and 0.073 , respectively. Scattering factors for $\mathrm{C}, \mathrm{H}, \mathrm{N}$ and O were taken from ref. $8 c$. Molecular-geometry calculations were carried out by using the computer program PARST ${ }^{8 d}$ and the structure drawing by using the ORTEP program. ${ }^{8 e}$

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